

**Solution**

**FULL TEST 1 PHYSICS**

**NEET-UG - Physics**

1. (a) moment of force

**Explanation:** [Moment of force] = [Torque] =  $[ML^2T^{-2}]$

2. (a) 20.0

**Explanation:** Initial velocity,  $u = 100 \text{ m/s}$

As it stops so final velocity,  $v = 0$

Acceleration  $a = -5 \text{ m/s}^2$

We know,  $v - u = at$

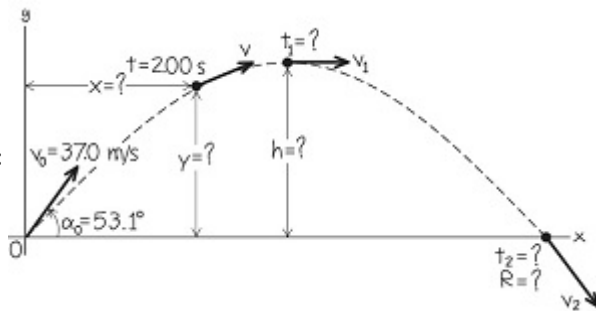
$$\Rightarrow t = \frac{v-u}{a}$$

$$\Rightarrow t = \frac{0-100}{-5} = \frac{-100}{-5}$$

$$\Rightarrow t = 20.0 \text{ s}$$

3. (a) 3.02 s, 44.7 m

**Explanation:**



The initial velocity of the ball has components

$$v_{0x} = v_0 \cos \alpha_0 = 37.0 \times \cos 53.1^\circ$$

$$= 22.2 \text{ m/s}$$

$$v_{0y} = v_0 \sin \alpha_0 = 37.0 \times \sin 53.1^\circ$$

$$= 29.6 \text{ m/s}$$

At the highest point, the vertical velocity  $v_y$  is zero. Call the time when this happens  $t_1$ ; then

$$v_y = v_{0y} - gt_1 = 0$$

$$\Rightarrow t_1 = \frac{v_{0y}}{g} = \frac{29.6}{9.8} = 3.02 \text{ s}$$

The height at the highest point is the value of  $y$  at time  $t_1$ :

$$h = v_{0y}t_1 - \frac{1}{2}g(t_1)^2$$

$$= 29.6 \times 3.02 - \frac{1}{2} \times 9.8 \times (3.02)^2$$

$$= 44.7 \text{ m}$$

4. (a) Both A and R are true and R is the correct explanation of A.

**Explanation:** If  $\theta$  be the angle between two vectors  $\vec{A}$  and  $\vec{B}$ , then their scalar product,  $\vec{A} \cdot \vec{B} = AB \cos \theta$ . If  $\theta = 90^\circ$  then  $\cos \theta = 0$

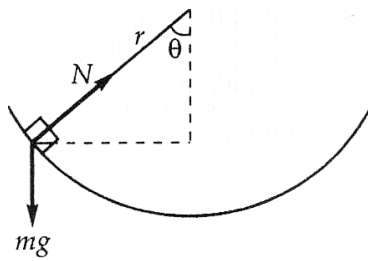
$$\therefore \vec{A} \cdot \vec{B} = 0$$

i.e., if  $\vec{A}$  and  $\vec{B}$  are perpendicular to each other than their scalar product will be zero.

- 5.

(d)  $\omega = \sqrt{\frac{g}{r} \cos \theta}$

**Explanation:**



Resolving the forces along horizontal and vertical directions,

$$N \sin \theta = m\omega^2 r \sin \theta$$

$$\Rightarrow N = m\omega^2 r$$

$$\text{and } N \cos \theta = mg$$

On dividing, we get

$$\frac{1}{\cos \theta} = \frac{\omega^2 r}{g} \text{ or } \omega = \sqrt{\frac{g}{r \cos \theta}}.$$

6. (a) 0.25 m/s

**Explanation:** By conservation of momentum,

$$100 \times v = 0.25 \times 100$$

$$v = 0.25 \text{ m/s}$$

7.

- (b) 300 J

**Explanation:**  $U_1 = \frac{1}{2} kx_1^2 = 100 \text{ J}$

$$U_2 = \frac{1}{2} kx_2^2$$

$$\frac{U_2}{U_1} = \frac{x_2^2}{x_1^2} = \left(\frac{4}{2}\right)^2 = 4$$

$$U_2 = 4U_1 = 400 \text{ J}$$

$$\therefore U_2 - U_1 = 300 \text{ J}$$

8. (a) Both A and R are true and R is the correct explanation of A.

**Explanation:** When the water is at the top of the fall it has potential energy  $mgh$  (when  $m$  is the mass of the water and  $h$  is the height of the fall). On falling this potential energy is converted into kinetic energy, which further converted into heat energy and so temperature of water increases.

9.

- (d) 5 : 7

**Explanation:**  $E_{\text{tran}} = \frac{1}{2}mv^2$

$$E_{\text{rot}} = \frac{1}{2}I\omega^2 = \frac{1}{2} \times \frac{2}{5}mr^2 \times \omega^2 = \frac{1}{5}mv^2$$

$$E_{\text{tot}} = E_{\text{tran}} + E_{\text{rot}} = \left(\frac{1}{2} + \frac{1}{5}\right)mv^2 = \frac{7}{10}mv^2$$

$$\frac{E_{\text{tran}}}{E_{\text{total}}} = \frac{\frac{1}{2}mv^2}{\frac{7}{10}mv^2} = \frac{5}{7} = 5 : 7$$

10.

- (c) A is true but R is false.

**Explanation:** Torque = Force  $\times$  perpendicular distance of line of force from the axis of rotation. Hence for a given applied force, torque or true tendency of rotation will be high for large value of  $d$ . If distance  $d$  is smaller, then greater force is required to cause the same torque, hence it is harder to open or shut down the door by applying a force near the hinge.

11. (a) Both  $mg$  and  $\frac{GmM_e}{R_e^2}$

**Explanation:** Force on particle at surface is

$$F = mg$$

where,  $g$  = acceleration due to gravity at the earth's surface

$$\text{Also, } g = \frac{GM_e}{R_e^2}$$

$$\Rightarrow F = mg = \frac{GmM_e}{R_e^2}$$

12. (a) Assertion and reason both are correct statements and reason is correct explanation for assertion.

**Explanation:** Assertion and reason both are correct statements and reason is correct explanation for assertion.

13.

(b) 1

**Explanation:** Strain =  $\frac{\text{Change in length}}{\text{Original length}} = \frac{2L-L}{L} = 1$

14.

(c) 2 : 1

**Explanation:**  $Y = \frac{Fl}{A\Delta l}$

$$\Rightarrow \Delta l = \frac{Fl}{AY}$$

$$(\Delta l)_{\text{steel}} = (\Delta l)_{\text{Brass}}$$

$$\text{or } \frac{W_S l}{AY_S} = \frac{W_B l}{AY_B}$$

$$\therefore \frac{W_S}{W_B} = \frac{Y_S}{Y_B} = \frac{2}{1} = 2 : 1$$

15.

(d) size of orifice

**Explanation:** Velocity of efflux,  $v = \sqrt{2gh}$

Clearly, it does not depend on the size of the orifice.

16.

(d)  $2\sqrt{h(H-h)}$

**Explanation:** From the figure, it can be considered that the water falls from a height  $H - h$ . For the fall of the water from such height, the equation is

$$H - h = \frac{1}{2}gt^2$$

Here, time taken by the water to reach the ground is  $t$ .

From the above equation, the time taken is

$$t = \sqrt{\frac{2(H-h)}{g}}$$

The horizontal distance covered by the water in time  $t$  from the foot of the tank is

$$x = vt$$

Insert  $t = \sqrt{\frac{2(H-h)}{g}}$  and  $v = \sqrt{2gh}$  in the above equation gives

$$x = vt$$

$$= (\sqrt{2gh}) \left( \sqrt{\frac{2(H-h)}{g}} \right)$$

$$= 2\sqrt{(H-h)h}$$

Therefore, the distance of the water from the foot of the wall is  $2\sqrt{(H-h)h}$ .

17.

(b) 384.8 K

**Explanation:** It is given that:

$$R = R_0 [1 + \alpha (T - T_0)] \dots(i)$$

where,

$R_0$  and  $T_0$  are the initial resistance and temperature respectively

$R$  and  $T$  are the final resistance and temperature respectively

$\alpha$  is a constant

At the triple point of water,  $T_0 = 273.15$  K

Resistance of lead,  $R_0 = 101.6 \Omega$

At normal melting point of lead,  $T = 600.5$  K

The resistance of lead,  $R = 165.5 \Omega$

Substituting these values in equation (i), we get:

$$R = R_0 [1 + \alpha (T - T_0)]$$

$$165.5 = 101.6 [1 + \alpha(600.5 - 273.15)]$$

$$1.629 = 1 + \alpha(327.35)$$

$$\therefore \alpha = \frac{0.629}{327.35} = 1.92 \times 10^{-3} \text{ K}^{-1}$$

For resistance,  $R_1 = 123.4 \Omega$

$$R_t = R_0 [1 + \alpha (T - T_0)]$$

where,

$T$  is the temperature when the resistance of lead is  $123.4 \Omega$

$$123.4 = 101.6 [1 + 1.92 \times 10^{-3} (T - 273.15)]$$

Solving for  $T$ , we get

$$T = 384.8 \text{ K.}$$

18.

$$(c) \frac{R^2 \sigma (t+273)^4}{r^2}$$

**Explanation:** From Stefan's law, energy radiated by the sun per second,

$$E = \sigma AT^4 = \sigma \times 4\pi R^2 \times T^4$$

Power received per unit area at distance  $r$  from the sun,

$$I = \frac{E}{4\pi r^2} = \frac{\sigma \times 4\pi R^2 \times T^4}{4\pi r^2} = \frac{\sigma R^2 (t+273)^4}{r^2}$$

19.

(d) 0.5 atm

**Explanation:**  $PV = P_f \cdot 2V$

$$P_f = \frac{V}{2V}$$

$$P_f = 0.5P$$

20.

(c) Assertion is correct statement but reason is wrong statement.

**Explanation:** Assertion is correct statement but reason is wrong statement.

21.

$$(c) P = \frac{2E}{3}$$

**Explanation:**  $P = \frac{2E}{3}$

22.

(d) 50 cm/sec

**Explanation:** Given  $y = 0.25 \sin(200t)$

compare this equation with the standard equation

$$y = A \sin(\omega t)$$

$$\text{amplitude } A = 0.25 \text{ cm}$$

$$\omega = 200$$

$$v_{\max} = \omega A$$

$$v_{\max} = 200 \times 0.25$$

$$v_{\max} = 50 \text{ cm/s}$$

23.

(a) Second

**Explanation:** Frequencies that are an integral multiple of the fundamental frequency are called harmonics or overtones.

$$f_n = \frac{nv}{2L}$$

$n = 1$ , is called the fundamental frequency

$n = 2$ , 1st overtone or 2nd harmonic

$n = 3$ , 2nd overtone or 3rd harmonic

24.

(b) 2

**Explanation:**  $F = \frac{1}{4\pi\epsilon_0} \frac{q(Q-q)}{x^2}$

$x$  is constant. For maximum force,

$$\frac{dF}{dq} = 0$$

$$\Rightarrow \frac{d}{dq} (qQ - q^2) = 0$$

$$\therefore Q - 2q = 0 \Rightarrow \frac{Q}{q} = 2$$

25. (a)  $\frac{1}{6} \frac{4\pi q}{4\pi\epsilon_0}$

**Explanation:**  $\phi_E = \frac{q}{\epsilon_0} = \frac{1}{6} \frac{4\pi q}{(4\pi\epsilon_0)}$

26.

(b) 400 V

**Explanation:**  $V = \frac{C_1 V_1 + C_2 V_2}{C_1 + C_2}$   
 $= \frac{20\mu\text{F} \times 500 \text{ V} + 10\mu\text{F} \times 200 \text{ V}}{20\mu\text{F} + 10\mu\text{F}}$   
 $= \frac{12000}{30} \text{ V} = 400 \text{ V}$

27.

(b) Assertion and reason both are correct statements but reason is not correct explanation for assertion.

**Explanation:** Assertion and reason both are correct statements but reason is not correct explanation for assertion.

28.

(c) 10%

**Explanation:** Let original Current In lamp = I

Resistance of Lamp = R

Then power P = I<sup>2</sup>R

According to question,

New Current  $I_n = I - I \times \frac{5}{100} = \frac{19}{20} I$

Resistance = R

New power  $P_n = I_n^2 R = \left(\frac{19}{20} I\right)^2 R = \frac{361}{400} I^2 R$

Power decrease =  $I^2 R - \frac{361}{400} I^2 R = \frac{39}{400} I^2 R$

% Decrease =  $\frac{\text{change in power}}{\text{original power}} \times 100$

$= \frac{\frac{39}{400} I^2 R}{I^2 R} \times 100 = \frac{39 I^2 R}{400 I^2 R} \times 100$

$= \frac{39}{4} = 9.75\% \approx 10\%$

29. (a) along with a very slow net motion in the opposite direction of the field

**Explanation:** The electrons in a conductor have random velocities and when an electric field is applied, they suffer repeated collisions and in the process move with a small average velocity, opposite to the direction of the field. This is equivalent to positive charge flowing in the direction of the field.

30. (a) 250 Ω

**Explanation:**  $R_g = \frac{I_S}{V_S}$   
 $= \frac{5 \text{ div /mA}}{20 \text{ div /V}}$   
 $= \frac{5 \text{ V}}{20 \times 10^{-3} \text{ A}} = 250\Omega$

31.

(d) 5 mA

**Explanation:** 5 mA

32.

(c) diamagnetism

**Explanation:** Diamagnetism is a universal property among all substances.

33.

(c) a torque but not a force

**Explanation:** a torque but not a force

34.

(c) Electromagnetic induction

**Explanation:** Electromagnetic induction, the electric dynamo uses rotating coils of wire and magnetic fields to convert mechanical rotation into a pulsing direct electric current through Faraday's law of induction.

35. (a) 0

**Explanation:** Power in an ac circuit,

$P = V_{rms} I_{rms} \cos \phi$

If only inductor is present in circuit, then voltage leads the current by  $90^\circ$  i.e.  $\phi = 90^\circ$

Hence,  $P = 0$  as  $\cos 90^\circ = 0$

36.

(d) zero

**Explanation:** Here, the phase difference between current and e.m.f.,

$$\phi = \pi/2$$

$$\therefore P_{av} = E_v I_v \cos \phi = E_v I_v \cos \pi/2 = 0$$

37.

(b)  $2.50 \times 10^{-6}$  N

$$\text{Explanation: } F_{av} = \frac{2IA}{c} = \frac{2 \times 25 \times 10^4 \times 15 \times 10^{-4}}{3 \times 10^8} \text{ N}$$

$$= 250 \times 10^{-8} \text{ N} = 2.5 \times 10^{-6} \text{ N}$$

38.

(a)  $\lambda_m > \lambda_v > \lambda_x$

**Explanation:** Since of the given regions, wavelength of microwave is highest and that of x-ray is minimum.

39.

(b) 10 s

**Explanation:** Exposure time,  $t \propto d^2$

$$\therefore t_2 = \frac{d_2^2}{d_1^2} t_1 = \frac{120^2}{60^2} \times 2.5 = 10 \text{ s}$$

40.

(a) its wavelength decreases

**Explanation:** The energy of the light is related to the frequency. When the light enters the medium, the apparent speed of light changes. If the frequency changed, the energy would not be conserved. The wavelength changes to balance the change in speed. When light enters from air to glass (from rarer to denser medium), its speed decreases as a consequence its wavelength also decreases.

41.

(d) Rectilinear motion

**Explanation:** Newton's corpuscular theory of light is based on the following points:

1. Light consists of very tiny particles known as "corpuscles".
2. These corpuscles on emission from the source of light travel in a straight line with high velocity.
3. These particles are emitted from a source of light in all directions.

42.

(b) 1.78

**Explanation:** Position of 8th bright fringe in the medium = Position of 5th dark fringe in air

$$\frac{8\lambda D}{d} = (2 \times 5 - 1) \frac{\lambda D}{2d}$$

$$\Rightarrow \frac{8\lambda D}{\mu d} = \frac{9\lambda D}{2d}$$

$$\therefore \mu = \frac{16}{9} = 1.78$$

43.

(b) momentum

**Explanation:** As  $p = \frac{h}{\lambda}$ , so electrons and photons having the same wavelength  $\lambda$  will have the same momentum  $p$ .

44.

(b) zero

**Explanation:** zero

45.

(a)  $n\lambda$

**Explanation:** The circumference of the orbit must be an integral multiple of de Broglie wavelength  $\lambda$

$$\text{i.e.; } 2\pi r = n\lambda$$

46.

(c) The helium nucleus has more kinetic energy than the thorium nucleus.

**Explanation:**  ${}_{92}^{238}\text{U} \rightarrow {}_{90}^{214}\text{Th} + {}_2^4\text{He}$

By conservation of linear momentum,

$$p_f = p_i = 0$$

$$\therefore p_{He} - p_{Th} = 0 \Rightarrow p_{He} = p_{Th}$$
$$\Rightarrow K_{He} > K_{Th}$$

47. (a) more than dc value

**Explanation:** The output of a rectifier consists of an AC component called ripple and a DC component. The ratio of the AC and DC components will be greater than one for a half-wave rectifier, while it will be less than one for a full-wave rectifier.

48.

(b) Forward biasing, 0 A

**Explanation:** Forward biasing, 0 A

49.

(c) added in

**Explanation:** added in

50. (a) curve C

**Explanation:** curve C

JUPITER

**Solution**  
**MODEL QUESTION PAPER 1**  
**NEET-UG - Chemistry**

1. (a) Both Normality and Molarity

**Explanation:**

Since Molarity and Normality both the modes of expression to represent the strength of a solution are volume-dependent, a variation in temperature of the solution results in a change in concentration or strength of the solution. It is due to the thermal expansion of liquids.

2.

- (c) 0.03 mL

**Explanation:**

Since, 1 microliter

$$= 1 \times 10^{-6} \text{ liters, \& 1ml}$$

$$= 1 \times 10^{-3}$$

$\therefore$  1 microliter

$$= 0.001 \text{ mL}$$

$\therefore$  30 microliters

$$= (0.001 \times 30) \text{ mL}$$

$$1 \text{ microliter} = 0.03 \text{ milliliter}$$

3.

- (c) electrons

**Explanation:**

Cathode rays - In 1897, British physicist J. J. Thomson showed the rays were composed of a previously unknown negatively charged particle, which was later named the electron.

4.

- (d) Balmer series

**Explanation:**

Balmer series

5.

- (c) Group-1 (alkali metals) and Group-2 (alkaline earth metals)

**Explanation:**

S-block elements comprise of Group 1 (alkali metals) and Group 2 (alkaline earth metals) with the general outermost electronic configuration as  $ns^{1-2}$ .

6.

- (b) principal quantum number

**Explanation:**

In the modern periodic table, each period indicates the value of principal quantum number (n). It also implies the number of shells or orbits.

7. (a) SF<sub>2</sub>O

**Explanation:**

SF<sub>2</sub>O has pyramidal shape as it is sp<sup>3</sup> hybridised and it has 3bp and 1 lp.

8. (a) 3,3,3

**Explanation:**



Total number of electrons in the  $N_2$  molecule is  $7+7=14$ .

As per the formula Bonded pair of electrons  $N_b$ :  $\sigma 1s^2 \sigma 2s^2 \pi 2p_y^2 \pi 2p_z^2 \sigma 2p_x^2$

Total 10 electrons.

Anti bond pairs of electrons  $N_a$ :  $\sigma 1s^*^2 \sigma 2s^*^2$  Total 4 electrons.

$$\text{Bond Order (B.O.)} = \frac{N_b - N_a}{2} = \frac{10 - 4}{2} = 3$$

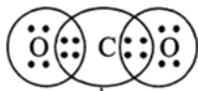
Similarly, the bond order of CO and  $NO^+$  is 3.

9.

(c)  $CO_2$

**Explanation:**

Lewis structure of  $CO_2$  can be represented as



Complete octet

10. (a)  $-5.744 \text{ kJ mol}^{-1}$

**Explanation:**

$$\Delta_r G^0 = -RT \ln K_c$$

$$\Delta G^0 = -2.303 \times 8.314 \times 300 \times \log 10 = -5744.14 \text{ Jmol}^{-1} = -5.74414 \text{ kJmol}^{-1}$$

11.

(b) zero

**Explanation:**

By convention, the standard enthalpies of formation of all elements in their most stable states (reference states) are zero.

12.

(d)  $1.8 \times 10^{-3}$

**Explanation:**

$$K_c = \frac{[PCl_3][Cl_2]}{[PCl_5]} = 1.2 \times 10^{-3} \times 1.2 \times 10^{-3} / 0.8 \times 10^{-3} = 1.8 \times 10^{-3} \text{ L mol}^{-1}$$

13.

(b)  $P_{CO_2}$  and  $P_{CO} = 0.461 \text{ atm}$  and  $1.739 \text{ atm}$

**Explanation:**

For the given reaction;  $FeO + CO(g) = Fe(s) + CO_2(g)$

$$Q_p = \frac{P_{CO_2}}{P_{CO}} = \frac{0.80}{1.4} = 0.571$$

It is given that  $K_p = 0.265$

Since  $Q_p > K_p$

The reaction will proceed in the backward reaction.

Therefore, we can say that the pressure of CO will increase while the pressure of  $CO_2$  will decrease.

Now, let the increase in pressure of CO = decrease in pressure of  $CO_2$  be P.

Then we can write

$$K_p = \frac{P_{CO_2}}{P_{CO}} = \frac{0.80 - P}{1.4 + P}$$

$$\Rightarrow 0.265 = \frac{0.80 - P}{1.4 + P}$$

$$\Rightarrow 0.265(1.4 + P) = 0.80 - P$$

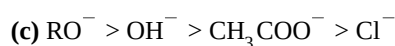
$$\Rightarrow 0.371 + 0.265P = 0.80 - P$$

$$\Rightarrow 1.265P = 0.429 \Rightarrow P = 0.339 \text{ atm}$$

Therefore, equilibrium partial pressure of  $CO_2 = 0.80 - 0.339 = 0.461 \text{ atm}$

Also, equilibrium partial pressure of CO =  $1.4 + 0.339 = 1.739 \text{ atm}$

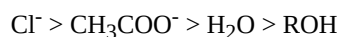
14.



**Explanation:**

conjugate acids of given bases are  $\text{H}_2\text{O}$ ,  $\text{ROH}$ ,  $\text{CH}_3\text{COOH}$ ,  $\text{HCl}$

their acidic strength in the order



basic strength in the order  $\text{RO}^- > \text{OH}^- > \text{CH}_3\text{COO}^- > \text{Cl}^-$

15.

(b) 0.04

**Explanation:**

The second equation can be obtained by reversing the 1st reaction.

$$K_1 = \frac{[\text{HI}]}{[\text{H}_2]^{1/2}} \times [I_2]^{1/2}$$

square the equation

$$K_1^2 = \frac{[\text{HI}]^2}{[\text{H}_2]} \times [I_2] \dots\dots\dots (1)$$

$$K_2 = [\text{H}_2] \times [I_2] / [\text{HI}] \dots\dots\dots (2)$$

From eq (1) and (2)

$$(K_1)^2 = 1/K_2$$

$$(5.2)^2 = 1/K_2$$

$$K_2 = 1/27.04 = 0.0369$$

16.

(c)  $\text{H}^+$  ions

**Explanation:**

If a reaction is carried out in acidic medium,  $\text{H}^+$  ions are used to balance the equation. If it is carried out in basic medium,  $\text{OH}^-$  ions are used.

17.

(b)  $\text{BF}_3$

**Explanation:**

$\text{BF}_3$  is Lewis acid and also used as a catalyst in some reaction e.g. it initiates polymerisation reactions.

18.

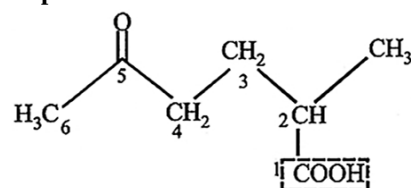
(c) Covalent bond

**Explanation:**

A covalent bond is formed between carbon and nitrogen atom. If duplet (2) or octet (8) is completed by sharing of electrons between two electronegative elements, the bond formed is called covalent bond.

19. (a) 2-Methyl-5-oxohexanoic acid

**Explanation:**



IUPAC NAME

2-Methyl-5-oxohexanoic acid

20. (a) - COOR (R = alkyl group right), - COCl, - CONH<sub>2</sub>, - CN, - HC = O, > C = O, - OH, - H<sub>2</sub>

**Explanation:**

COOR (R=alkyl group) > -COCl, > -CONH<sub>2</sub> > -CN > -HC=O > >C=O > -OH, -NH<sub>2</sub> is the decreasing order of priority  
ester > acyl halide > acyl amide > nitrile > aldehyde > ketones > alcohol > amine

21.

(b) P > R > Q

**Explanation:**

8 $\alpha$ -H in P.

22.

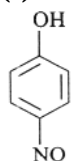
(c) A > B > C > D

**Explanation:**

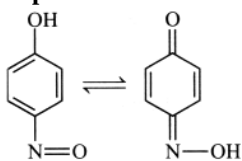
C<sub>2</sub>H<sub>5</sub>SH is more acidic than C<sub>2</sub>H<sub>5</sub>OH as S — H bond is weaker than O — H bond.

23.

(c)



**Explanation:**



24.

(c) 3,6 - Diethyl - 2 - methyloctane

**Explanation:**

Following the rules of nomenclature, the IUPAC name of the given compound is 3, 6-Diethyl-2-methyloctane.

25.

(b) one C - C  $\sigma$  bond, two C - H  $\sigma$  bonds and two C - C  $\pi$  bonds.

**Explanation:**

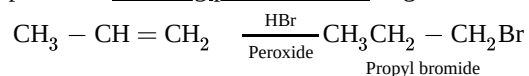
Ethyne is an unsaturated compound which belongs to the alkyne family and has the formula C<sub>2</sub>H<sub>2</sub>. Thus it consists of one C—C  $\sigma$  bond, two C—H  $\sigma$  bonds, and two C—C  $\pi$  bonds.

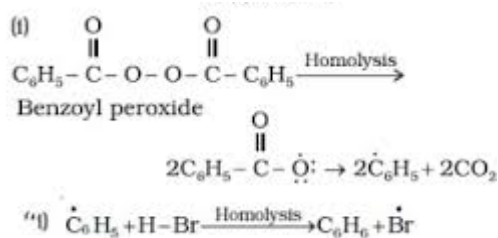
26.

(b) a free radical mechanism.

**Explanation:**

The mechanism involved along with the generation of a free radical [*i. e. phenyl free radical*(C<sub>6</sub>H<sub>5</sub>)] by the use of benzoyl peroxide, showing peroxide effect in given below,



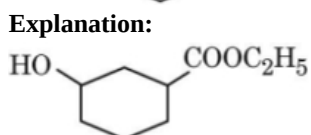
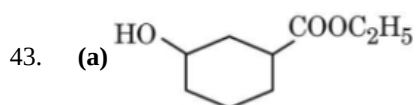


27. (c) proportional to the pressure of the gas over the liquid  
**Explanation:**  
 For the dissolution of gases in liquids, Henry's Law is applicable i.e. the mass of the gas dissolved in a given solvent at any temperature is proportional to the pressure of the gas above the solvent.
28. (c)  $\text{HNO}_3 + \text{H}_2\text{O}$   
**Explanation:**  
 Negatively deviated non ideal solution.
29. (c) cell emf  
**Explanation:**  
 cell emf
30. (d) Lead- storage battery cell  
**Explanation:**  
 Lead storage battery is a secondary cell while leclanche cell and mercury cell are examples of primary batteries.
31. (a) infinite  
**Explanation:**  
 The reaction would be 100 % complete only after infinite time which cannot be calculated.
32. (d) four times  
**Explanation:**  
 four times
33. (c) KI  
**Explanation:**  
 KI reacts with  $\text{I}_2$  to form  $\text{I}_3^-$ . The solubility of Iodine in water in the presence of KI due to the formation of potassium triiodide.  $\text{KI}_3$  easily breaks down and thus the solution has all the properties of free iodine.  

$$\text{KI} + \text{I}_2 \rightleftharpoons \text{KI}_3$$
34. (d) 12  
**Explanation:**  

$$2\text{KClO}_3 \rightarrow 2\text{KCl} + 3\text{O}_2$$
  
 2mol of  $\text{KClO}_3$  gives 3 mol of  $\text{O}_3$ .  
 So 8 mol of potassium chlorate will yield =  $\frac{8 \times 3}{2} = 12$  mol of  $\text{O}_2$ .

35. **(b) 26**  
**Explanation:**  
 $[\text{Ar}]3d^5$ : As  $X^{3+}$  is formed by the loss of 3 electrons, the configuration of element X is  $[\text{Ar}]3d^64s^2$   
 $\therefore$  Atomic number = 26
36. **(a)  $3d^54s^2$**   
**Explanation:**  
 $3d^54s^2$  shows the highest oxidation state
37. **(b) 4**  
**Explanation:**  
Cobalt atoms have an electronic configuration of  $4s^2 3d^7$  but its electronic configuration becomes  $4s^0 3d^6$  after losing three electrons. Thus  $\text{Co}^{3+}$  ions have 4 unpaired electrons present in 3d orbital.
38. **(d)  $[\text{Cu}(\text{NH}_3)_4]^{+2}$**   
**Explanation:**  
Complexes in which the central metal is bound to only one kind of donor groups are called homoleptic complexes.  $[\text{Cu}(\text{NH}_3)_6]^{+2}$  is a homoleptic complex because in this only ammonia group is the donor group bound to  $\text{Cu}^{+2}$ .
39. **(a) Ethanol**  
**Explanation:**  
Ethanol
40. **(a)  $\text{SOCl}_2$  in presence of pyridine**  
**Explanation:**  
The hydroxyl group of an alcohol is replaced by halogen on reaction with concentrated halogen acids, phosphorus halides, or thionyl chloride. Thionyl chloride ( $\text{SOCl}_2$ ) is preferred because the other two products  $\text{SO}_2$  and  $\text{HCl}$  are escapable gases. Hence, the reaction gives pure alkyl halides.  
 $\text{ROH} + \text{SOCl}_2 \rightarrow \text{RCl} + \text{SO}_2(\text{g}) + \text{HCl}(\text{g})$
41. **(c) Williamson synthesis**  
**Explanation:**  
Williamson synthesis
42. **(a) Propan-1-ol, butan-2-ol, butan-1-ol, pentan-1-ol**  
**Explanation:**  
The boiling point increases with an increase in the molecular mass of the alcohol. Thus, the correct order is Propan-1-ol < butan-2-ol < butan-1-ol < pentan-1-ol.



44. (c)  $I_2$  and NaOH  
**Explanation:**  
 $I_2$  and NaOH
45. (d) N-Ethyl-N-methylethanamine  
**Explanation:**  
 By alkylation reaction of methyl amine with two moles of ethyl chloride, N-Ethyl-N-methylethanamine is formed.
- $$CH_3NH_2 + C_2H_5Cl \longrightarrow CH_3 - \underset{\substack{| \\ CH_2CH_3}}{NH} \xrightarrow{C_2H_5Cl} CH_3 - \underset{\substack{| \\ C_2H_5}}{N} - C_2H_5$$
- (N-ethyl-N-methylethanamine)
46. (b) Aluminium chloride reacts with Aniline  
**Explanation:**  
 $AlCl_3$  being a Lewis acid reacts with the lone pair of  $-NH_2$  group of aniline forming an adduct ( $C_6H_5NH_2^+ AlCl_3^-$ ) which deactivates the benzene system hence no Friedel-Crafts reaction occurs.
47. (b) It gives 2, 4 DNP test.  
**Explanation:**  
 Glucose doesn't react with Schiff's reagent & 2,4 DNP reagent despite having an aldehyde group because there is no free aldehyde group present in the cyclic form as it forms hemiacetal. Thus, it does not give 2,4-DNP test.
48. (a) preparing the uterus for implantation of fertilised egg.  
**Explanation:**  
 The combination of ovum and sperm forms the fertilized egg which further gets implanted in the uterus. This is brought about by the hormone progesterone which prepares uterus for this process.
49. (a) Carius tube  
**Explanation:**  
 Tube in which a known mass of an organic compound is heated for the quantitative analysis of S is known as Carius tube.
50. (b) 1-chloro-2, 4-dinitrobenzene  
**Explanation:**  
 From the given information, the organic compound must contain N and Cl atoms. Hence, the organic compound would be 1-chloro-2, 4-dinitrobenzene.

**Solution**  
**BOTANY MODEL PAPER 1**  
**NEET-UG - Biology**  
**BOTANY (Section-A)**

1. **(a)** Name of the author

**Explanation:**

In binomial nomenclature, the name of the author appears after the species name, i.e., at the end of the biological name and is written in an abbreviated form or in full. *Mangifera indica* Linn. indicates that this species was first described by Linnaeus. This method of mentioning the author's name is called a citation.

2.

**(b)** *Triticum aestivum*

**Explanation:**

Biological name of wheat also called bread wheat is *Triticum aestivum*.

3.

**(b)** All of them have helical symmetry

**Explanation:**

Self-assembly of virus capsids follows two basic patterns: helical symmetry, in which the protein subunits and the nucleic acid are arranged in a helix, and icosahedral symmetry, in which the protein subunits assemble into a symmetric shell that covers the nucleic acid-containing core.

4.

**(d)** T.O. Diener

**Explanation:**

In 1971, T.O. Diener discovered a new infectious agent that was smaller than viruses and caused potato spindle tuber disease. It was found to be a free RNA; it lacked the protein coat that is found in viruses, hence the name viroid. The RNA of the viroid was of low molecular weight.

5.

**(b)** Pericarp

**Explanation:**

Pericarp

6.

**(b)** One fuses with the egg and other fuses with central cell nuclei

**Explanation:**

In flowering plants,

- Out of the two male gametes, one gamete fuses with the nucleus of the egg cell and forms the zygote. The process is known as syngamy.
- The other male gamete fuses with the two polar nuclei located in the central cell to form a triploid primary endosperm nucleus (PEN). Since the process involves the fusion of three haploid nuclei, it is known as triple fusion.
- Since two kinds of fusions (syngamy and triple fusion) take place in an embryo sac it is known as double fertilisation.

7.

**(c)** *Funaria*

**Explanation:**

*Funaria* is a moss plant and the predominant stage of the life cycle of a moss is the gametophyte which consists of two stages. The first stage is the protonema stage, which develops directly from a spore. It is a creeping, green, branched, and frequently filamentous stage.

8. **(c)** C, F, and E  
**Explanation:**  
The parts of the grass embryo labelled as C,F, and B represent the coleoptile, shoot apex, and coleorhiza, respectively.
9. **(c)** microsporophylls  
**Explanation:**  
Microsporophylls are the leaf-like structures that bear the microsporangia and stamens of the angiosperms bear the microsporangia in their anthers these represent the modified microsporophylls.
10. **(b)** Phloem parenchyma  
**Explanation:**  
The described statements are associated with the phloem parenchyma. Phloem parenchyma is living and has thin cell walls. These cells form the packing tissue between all the other types of cells. These cells store compounds such as starch.
11. **(a)** Exarch  
**Explanation:**  
This condition is seen in roots.
12. **(d)** Monocot stem  
**Explanation:**  
Sclerenchymatous hypodermis, a large number of scattered vascular bundles, and sclerenchymatous bundle sheath can be observed in monocot stem (Maize stem, Sunflower stem, Brinjal root, and Grasses root).
13. **(a)**  $3A + XXY$   
**Explanation:**  
 $3A + XXY$
14. **(a)** homologous  
**Explanation:**  
Non-disjunction is the non-separation of a pair of homologous chromosomes during anaphase I. This results in two cells with an extra copy of one chromosome and two cells without that chromosome.
15. **(c)** Base and Sugar  
**Explanation:**  
Base and Sugar
16. **(d)** DNA but no histones  
**Explanation:**  
DNA but no histones
17. **(b)** Fission  
**Explanation:**  
Fission



18. **(a)** 2 + 9 pattern  
**Explanation:**  
The arrangement of central and outer microtubules in a cilium is called the 2 + 9 pattern. Here 2 microtubules are in the center and 9 surrounding the center.
19.  
**(b)** Leukaemia  
**Explanation:**  
Leukaemia
20.  
**(c)** Health is a state of complete physical, mental, and social well-being.  
**Explanation:**  
Health could be defined as a state of complete physical, mental, and social well-being.
21.  
**(b)** are identical in size and appearance.  
**Explanation:**  
Homologous chromosomes are chromosomes having same structural features. In diploid nuclei, pairs of homologous chromosomes can be identified at the start of meiosis. One member of each pair comes from the female parent and other from the male parent. Homologous chromosomes have the same pattern of genes along the chromosome, but the nature of the genes may differ.
22. **(a)** (i) and (iii)  
**Explanation:**  
Thermoregulation is energetically expensive for many organisms particularly true for small animals like shrews and humming birds. During the course of evolution, the costs benefits of maintaining a constant internal environment are taken into consideration.
23.  
**(b)** More than one  
**Explanation:**  
Primary production is defined as the amount of biomass or organic matter produced per unit area over a time period by plants during photosynthesis. It is expressed in terms of weight ( $g^{-2}$ ) or energy ( $kcal\ m^{-2}$ ). The rate of biomass production is called productivity. It is expressed in terms of  $g^{-2}\ yr^{-1}$  or  $(kcal\ m^{-2})\ yr^{-1}$  to compare the productivity of different ecosystems.
24.  
**(d)** Fix atmospheric nitrogen  
**Explanation:**  
Fix atmospheric nitrogen
25.  
**(c)** All of these  
**Explanation:**  
A sacred grove or sacred wood is a grove of trees of special religious importance to a particular culture.
26.  
**(c)** Species area relationships  
**Explanation:**  
Species area relationships
27.  
**(b)** Tea

**Explanation:**

Oolong tea in Taiwan and brick tea in China are commercial varieties of tea. Dust is leftover small parts of leaves widely used.

28. (a) Tubulin

**Explanation:**

Spindle fibres are the cytoskeletal structure of eukaryotic cells, that are formed during cell division. It contracts and expands to separate sister chromatids between daughter cells. It is made up of microtubules, which are composed of 97% tubulin protein, and 3% RNA. The polymeric tubulin is formed by alpha and beta loop. In cilia, the microtubules are arranged in the 9+2 arrangement.

29.

(c) ABA

**Explanation:**

(ABA) Cis-**abscisic acid** decreases the rates of **cell** growth and **cell division** in the meristem.

30. (a) H<sub>2</sub>O

**Explanation:**

The excited electron from reaction centre of PS II is captured by a primary electron acceptor of the electron transport chain. The electron in the reaction centre needs to be replenished so as to repeat the cycle. This occurs by oxidation of water.

31.

(b) Mesophyll

**Explanation:**

Mesophyll is the site in plant leaves, where chlorophyll pigment is present. Thus carbon dioxide fixation occurs there, which produces malic acid. It is then transferred to agranal chloroplast in bundle sheath.

32. (a) CO<sub>2</sub> and light, respectively

**Explanation:**

There is a linear relationship between incident light and CO<sub>2</sub> fixation rates at low light intensities. At higher light intensities, gradually the rate does not show further increase as CO<sub>2</sub> become limiting factor.

33.

(c) 3: 1

**Explanation:**

The molar ratio of chlorophyll to xanthophyll is 3:1. Chlorophyll is a green pigment, present in all green plants and in cyanobacteria, which is responsible for the absorption of light to provide energy for photosynthesis. Xanthophyll is a yellow or brown carotenoid plant pigment which causes the autumn colours of leaves.

34. (a) Glyceraldehyde - 3 - phosphate to 1,3 - diphosphoglycerate

**Explanation:**

In glycolytic pathway, 3PGAL is converted into 1,3 - diphosphoglyceric acid by an oxidation and phosphorylation reaction, which occurs in the presence of H<sub>3</sub>PO<sub>4</sub> and coenzyme NAD.  $3\text{-phosphoglyceraldehyde} + \text{NAD}^+ + \text{Pi} \rightarrow 3\text{-phosphoglyceraldehyde dehydrogenase 1, 3 - phosphoglyceraldehyde acid} + \text{H}^+$

35.

(c) ABA

**Explanation:**

Abscisic acid (ABA) is formed from three isoprene subunits and is a key hormone in plant adaptation to environmental stresses. Abscisic acid (ABA) is often referred to as an inhibitory rather than stimulatory hormone. It is involved in the closure of stomata, bud, and seed dormancy and is known to inhibit other hormone actions. It promotes senescence and ageing in other parts of plants.

**BOTANY (Section-B)**

36. (a) Tautonyms do not occur in plants

**Explanation:**

Tautonyms are the species names in which specific epithet is the exact repeat of the generic name with/without transcribed symbol. The tautonyms are never accepted by botanical nomenclature as these are prohibited. Hence, 'Tautonyms do not occur in plants' is true statement.

37.

(d) (i), (ii), (v) and (vi)

**Explanation:**

- The asexual spores are generally not found, but vegetative reproduction by fragmentation is common.
- The sex organs are absent, but plasmogamy is brought about by the fusion of two vegetative or somatic cells of different strains or genotypes. The resultant structure is dikaryotic which ultimately gives rise to basidium.
- Karyogamy and meiosis take place in the basidium producing four basidiospores.

38. (a) Wind

**Explanation:**

In gymnosperms, the male and the female gametophytes do not have an independent free-living existence. They remain within the sporangia retained on the sporophytes. The pollen grain is released from the microsporangium. They are carried in air currents and come in contact with the opening of the ovules borne on megasporophylls.

39.

(d) 1-Thalamus, 2-Seed, 3-Endocarp, 4-Mesocarp

**Explanation:**

1-Thalamus, 2-Seed, 3-Endocarp, 4-Mesocarp

40.

(b) Corymb

**Explanation:**

Corymb

41.

(c) The alleles of gene controlling coat colour are codominant.

**Explanation:**

Coat colour in short-horned cattle is an example of codominance. If a cow with black coat is crossed with a bull with white coat, the F1 hybrids possess neither black nor white coat colour, but have roan coat colour, where black and white patches appear separately. This implies that the alleles controlling white and black colours expressed themselves independently when present together. Such alleles are called codominant alleles and the inheritance pattern is called codominance.

42.

(c) Mitochondria

**Explanation:**

Mitochondria

43. (a) This structure also found in eukaryotes

**Explanation:**

**Mesosome** which is formed by the extensions of the plasma membrane into the cell. These extensions are in the **form of vesicles, tubules and lamellae**. They help in cell wall formation, DNA replication and distribution to daughter cells. They also help in respiration, secretion processes, to increase the surface area of the plasma membrane and enzymatic content.

44.

(c) fuel wood

**Explanation:**

Biogas is a fuel gas, a mixture consisting of 60% methane (CH<sub>4</sub>) and of 40% CO<sub>2</sub>. It is a renewable energy resulting from biomass. Biogas typically refers to a mixture of different gases produced by the breakdown of organic matter in the absence of

oxygen. It can be a good substitute for fuel wood.

45.

**(b)** 19

**Explanation:**

During anaerobic respiration, only two molecules of ATP are produced by each molecule of glucose. So, 19 glucose molecules are required to produce 38 ATP molecules.

46. **(a)** All of these

**Explanation:**

All of these

47.

**(d)** Leaching

**Explanation:**

Detritivores (e.g., earthworm) break down detritus into smaller particles. This process is called fragmentation. By the process of leaching, water soluble inorganic nutrients go down into the soil horizon and get precipitated as unavailable salts. Bacterial and fungal enzymes degrade detritus into simpler inorganic substances. This process is called as catabolism. All above process are together called humification. The humus is further degraded by some microbes and release of inorganic nutrients occur by the process known as mineralisation.

48.

**(d)** Point B to C

**Explanation:**

Point B to C

49. **(a)** Cell elongation

**Explanation:**

Cell elongation

50.

**(c)** Violet and blue

**Explanation:**

Chlorophyll and carotenoid are important pigments of photosynthetic plants. These pigments can absorb light in the range of 700nm to 400 nm. Carotenoid pigment is present in the thylakoid and absorbs light violet and blue and reflects longer wavelength i.e., yellow, red and orange light.

**Solution**  
**ZOOLOGY MODEL PAPER 1**  
**NEET-UG - Biology**  
**ZOOLOGY (Section-A)**

1.  
**(d)** Statement d is incorrect  
**Explanation:**  
Sponges, cnidarians, flatworms, and ctenophorans do not have circulatory systems.
2. **(a)** Torpedo  
**Explanation:**  
Torpedo is a cartilaginous fish having electric organs.
3. **(d)** Only intracellular  
**Explanation:**  
Digestion in Sycon and other sponges is intracellular.
4. **(a)** Facilitate communication between adjoining cells by connecting the cytoplasm for rapid transfer of ions, small molecules and some large molecules  
**Explanation:**  
Gap junctions facilitate the cells to communicate with each other by connecting the cytoplasm of adjoining cells, for rapid transfer of ions, small molecules, and sometimes big molecules.
5. **(b)** Potassium urate  
**Explanation:**  
Cockroach excretes nitrogenous waste in the form of soluble potassium urate, which is liberated into the haemolymph. It is taken up by the cells lining the Malpighian tubules which facilitate the absorption of urate where the potassium urate reacts with water and carbon dioxide to form potassium hydrogen carbonate and uric acid. The potassium hydrogen carbonate is absorbed back and the uric acid is excreted.
6. **(d)** Both Alveolar epithelium and Endothelium blood capillaries  
**Explanation:**  
Outer covering of pulmonary alveoli is alveolar epithelium while that of blood capillaries is endothelium. Thus, both alveolar epithelium and endothelium of blood capillaries separate air in pulmonary alveoli from blood capillaries.
7. **(b)** Thyroid cartilage which is unpaired  
**Explanation:**  
Thyroid cartilage which is unpaired
8. **(d)** Inhaling tidal volume only  
**Explanation:**  
Inhaling tidal volume only
9. **(c)** (i), (ii), (iv), (vi), (iii), (v)  
**Explanation:**

The correct sequence to initiate inspiration is

(i), (ii), (iv),(vi), (iii), and (v).

10. **(d) Spirometer**  
**Explanation:**  
Spirometry is the process of recording the changes in the volume and movement of air in and out of the lungs and the instrument used for this purpose is called spirometer or respirometer, which measures the volume of air inhaled and exhaled by lungs.
11. **(b) Causes strong uterine contractions during parturition**  
**Explanation:**  
During parturition, signals originate from both the fully developed foetus and the placenta which induce mild uterine contractions called foetal ejection reflex. This triggers release of oxytocin from the maternal pituitary. Oxytocin acts on the uterine muscle and causes stronger uterine contractions.
12. **(a) Estrogen and progesterone**  
**Explanation:**  
The cycle of events starting from one menstruation till the next one is called the menstrual cycle. The menstrual cycle is complex controlled by the sex hormones - oestrogen and progesterone produced by ovaries.
13. **(c) Immunoglobulin A**  
**Explanation:**  
Colostrum, the yellowish fluid secreted by the mother during the initial days of lactation is very essential to impart immunity to the newborn infant because it contains immunoglobulin A. It will impart naturally acquired passive immunity to the new-born.
14. **(d) Both (ii) and (iv)**  
**Explanation:**  
Numerous children have been produced by in vitro fertilisation without any abnormality. Infertility is the inability to conceive or produce children even after 2 years of unprotected sexual co-habitation. Often the female is blamed for the couple being childless, but more often than not, the problem lies in the male partner.
15. **(d) All STD are completely curable**  
**Explanation:**  
All STD are completely curable
16. **(b) A - (iii), B - (iv), C - (ii), D - (i)**  
**Explanation:**  
When migration of a section of population to another place and population occurs, gene frequencies change in the original as well as in the new population. New genes/alleles are added to the new population and these are lost from the old population. This is called gene flow. If the same change occurs by chance, it is called genetic drift. Natural selection is a process in which heritable variations enabling better survival are enabled to reproduce and leave greater number of progeny. If the mutation affects a gene, it will result in a new version of that gene, a new allele.
17. **(c) Hugo de Vries, who worked on evening primrose.**  
**Explanation:**

Hugo de Vries, who worked on evening primrose.

18. (a) Metanephric

**Explanation:**

Metanephric kidney filters waste from the blood but excretes them to the outside through a pair of tubes, the ureters. In the embryo, the wastes are excreted directly into the amniotic fluid. The metanephric kidney is the final adult kidney of reptiles, birds, and mammals.

19.

(b) Cray fishes

**Explanation:**

Cray fishes

20.

(b) All of these

**Explanation:**

Renal autoregulation involves feedback mechanisms intrinsic to the kidney that causes either dilation or constriction in the afferent arteriole so as to counteract blood pressure changes and keeps a steady GFR. It can take place by renin angiotensin mechanism, juxta-glomerulus apparatus and vasopressin.

21. (a) Contraction of muscles after death

**Explanation:**

Contraction of muscles after death

22.

(d) Large blood vessels

**Explanation:**

All hollow organs like the intestine, stomach, and urinary bladder contains single-unit smooth muscles. Muscles of body parts that carry flow like air or blood are carrying multi-unit smooth muscles.; body parts like lung and large arteries show multi-unit muscles.

23.

(c) Hinge joint

**Explanation:**

In hinge joint movement is possible in one direction only. Joint of malleus and incus, knee joint, elbow joint, articulation joint of the lower jaw.

24.

(c) Medulla oblongata + cerebellum + pons

**Explanation:**

The brain is divided into three regions (i) Forebrain (ii) Midbrain and (iii) Hindbrain.

The forebrain is the largest part of the brain. It is the main thinking region. It is made up of cerebrum, hypothalamus and thalamus. The midbrain does not have any further divisions. The hindbrain consists of cerebellum, pons and medulla.

25. (a) Medulla oblongata

**Explanation:**

Hypothalamus is the region of the forebrain in the floor of the third ventricle, linked with the thalamus above and the pituitary gland below. It contains important control temperature (at 37°C), It functions as a centre for the integration of hormonal and autonomic nervous activity through its control of the pituitary secretions.

26. (a) scotopic vision and is the function of rods.

**Explanation:**

The rods contain the rhodopsin pigment that is highly sensitive to dim light. It is responsible for scotopic (twilight) vision.

27. **(d) Aldosterone**  
**Explanation:**  
Conn's syndrome occurs by the over-secretion of aldosterone by adrenal glands.
28. **(a) Adrenaline**  
**Explanation:**  
Adrenaline
29. **(c) Vasoconstriction**  
**Explanation:**  
The circulatory system promotes heat retention! .conservation through vasoconstriction of vessels in the skin. To retain heat, the body wants to divert blood flow away from exposed surfaces, such as the skin. Vasodilation results from a decrease in smooth muscle contraction. Smooth muscle is found within the tunica media of arterial vessels. Vasodilation produces an increase in surface area of the affected vessels.
30. **(c) All of these**  
**Explanation:**  
The QRS complex represents the depolarisation of the ventricles. The contraction starts shortly after Q and marks the beginning of the systole.
31. **(c) Ca**  
**Explanation:**  
Ca
32. **(a) Selectable marker**  
**Explanation:**  
Selectable marker
33. **(c) Prevention of the multiplication of bacteriophage in bacteria**  
**Explanation:**  
Restriction enzymes are found in bacteria and archaea and provide a defense mechanism against invading viruses. They cut both strands of DNA when certain foreign nucleotides are introduced in the cell and prevent the multiplication of viruses.
34. **(c) Bacillus thuringiensis**  
**Explanation:**  
Bacillus thuringiensis
35. **(a) Root**  
**Explanation:**  
Meloidogyne incognita is a nematode which infects the roots of the tobacco plants and causes a great reduction in the yield.
- ZOOLOGY (Section-B)**
36. **(c) Echinoderm**  
**Explanation:**  
Asterias (Star fish) is an example of Echinoderm.



37. **(d) Cockroach**  
**Explanation:**  
Cockroach
38. **(b) semilunar and tricuspid valve**  
**Explanation:**  
Three semilunar valves are located at the base of the pulmonary trunk and aorta. Tricuspid valve guards the opening between the right atrium and right ventricle.
39. **(a) more surface area for diffusion of gases.**  
**Explanation:**  
Air flows through the lung's complicated network of air-carrying tubes called bronchioles, which branch off the windpipe like the many branches on a tree trunk. Each branch ends in tiny sacs (or pockets) called alveoli, which allow oxygen and carbon dioxide to move between the lungs and bloodstream. These increase the surface area for respiratory exchange of gases tremendously.
40. **(c) Signal for parturition comes from fully developed foetus and placenta**  
**Explanation:**  
The signals for parturition originate from the fully developed foetus and the placenta, which induces mild uterine contractions called foetal ejection reflex.
41. **(c) (i) and (iv)**  
**Explanation:**  
(i) and (iv)
42. **(a) Random mating**  
**Explanation:**  
Random mating
43. **(b) Ammonia**  
**Explanation:**  
Ammonia
44. **(d) Tibia and fibula**  
**Explanation:**  
Tibia and tarsals
45. **(c) Acetylcholine**  
**Explanation:**  
Acetylcholine
46. **(d) All of these**  
**Explanation:**  
All of these

47. **(d)** Adrenal gland  
**Explanation:**  
Adrenal gland
48. **(c)** 'Lubb' sound at the beginning of systole  
**Explanation:**  
The first heart sound is called lub. It is associated with the closure of the tricuspid and bicuspid valves.
49. **(c)** Insertional inactivation of alpha-galactosidase in recombinant bacteria  
**Explanation:**  
Insertional inactivation of alpha-galactosidase in recombinant bacteria
50. **(a)** Retrovirus  
**Explanation:**  
Retrovirus is commonly used as a vector for introducing a DNA fragment in human lymphocytes as in gene therapy of ADA deficiency.